

## 有機化學實驗

# Friedel-Crafts Alkylation of Benzene

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## 原理

#### 1. Nucleophilic substitution Sn1 Sn2:

$$R-X + N\ddot{u} \rightarrow R-Nu + X$$

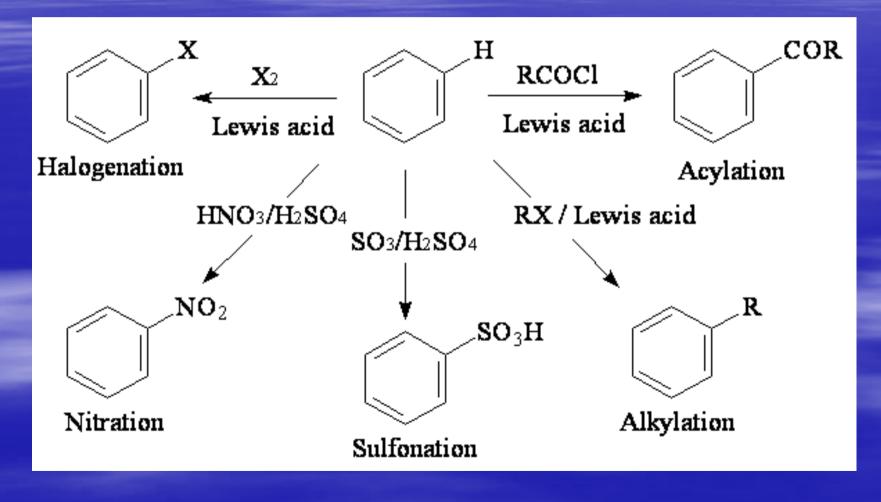
Electrophilic substitution SE : R-H + E<sup>+</sup> → R-E + H<sup>+</sup>

For aromatic system: Electrophilic aromatic substitution

#### 2. Mechanism:



#### **Reactions:**





#### Regioselectivity:

(ortho product)

(very stable)

- ※ (1) 畫出親電子取代基接在meta和para位置之中間體的所有共振圖形。
  - (2) 請解釋為何ortho 和para 的產物為主要產物?



### 合成步驟

#### **Reaction equation:**

1mL benzene + 2mL t-butyl chloride in 50mL R.B. flask (no water)

**↓(ice-water bath)** 

add 0.2g AICI3 and stir vigorously (in hood)

remove the flask from ice bath and stand for 5 min at r.t. (in hood)

add 3mL ice water to the mixture



concentration

transfer the mixture to a separatory funnel **↓(Add 3mL CH<sub>2</sub>Cl<sub>2</sub> to wash the flask)** extract the product with (2mL x 3) CH<sub>2</sub>Cl<sub>2</sub> wash the organic layer with 10mL dist. water **↓(collect the organic layer)** dry with MgSO4(anhy) filter and collect the filtrate



## 純化步驟

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the oily product + 5mL MeOH
heat to dissolve (water bath)
stand at r.t. until the crystal has produced
collect the crystal (wash it with the ice water)
suction to dry
weight
calculate the % yield (result report)
```



# Check out

1. 繳交產物並告知產物淨重。

2. 實驗問題: 2,4